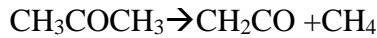


## ODE Practice Problems

### Nonisothermal Plug-Flow Reactor

Write general MATLAB functions for integrating simultaneous nonlinear differential equations using the Euler and Runge Kutta methods. Apply these functions for the solution of differential equations that simulate a nonisotherm plug reactor, as described below:

Vapor-phase cracking of acetone, described by the following endothermic reaction:



takes place in a jacketed tubular reactor. Pure acetone enters the reactor at a temperature of  $T_o=1035\text{K}$  and pressure of  $P_o=162\text{ kPa}$ , and the temperature of external gas in the heat exchanger is constant at  $T_a=1150\text{K}$ . Other data are as follows:

Volumetric flow rate:  $v_o=0.002\text{m}^3/\text{s}$

Volume of the reactor:  $V_R=1\text{m}^3$

Overall heat transfer coefficient:  $U=110\text{ W/m}^2\text{K}$

Heat transfer area:  $a=150\text{ m}^2/\text{m}^3\text{ reactor}$

Reaction constant:  $k=3.58 \exp[34222 * ((1/1035) - (1/T))] \text{ s}^{-1}$

Heat of reaction:

$$\Delta H_R = 80770 + 6.8(T-298) - 5.75 \times 10^{-3}(T^2-298^2) - 1.27 \times 10^{-6}(T^3-298^3) \text{ J/mol}$$

Heat capacity of acetone:  $C_{pA}=26.63 + 0.1830*T - 45.86 \times 10^{-6} T^2 \text{ J/mol.K}$

Heat capacity of ketene:  $C_{pB}=20.04 + 0.0945*T - 30.95 \times 10^{-6} T^2 \text{ J/mol.K}$

Heat capacity of methane:  $C_{pC}=13.39 + 0.077*T - 18.71 \times 10^{-6} T^2 \text{ J/mol.K}$

Determine the temperature profile of the gas along the length of the reactor. Assume constant pressure throughout the reactor.

**Method of Solution:** In order to calculate the temperature profile in the reactor, we have to solve the material balance and energy balance equations simultaneously:

Mole balance: 
$$\frac{dX}{dV} = \frac{-r_A}{F_{A0}}$$

Energy balance: 
$$\frac{dT}{dV} = \frac{Ua(T_a - T) + r_A \Delta H_R}{F_{A0}(C_{pA} + X \Delta C_p)}$$

where  $X$  is the conversion of acetone,  $V$  is the volume of the reactor,  $F_{A0}=C_{A0} * v_o$  is the molar flow rate of acetone at the inlet,  $T$  is the temperature of the reactor,

$\Delta C_p = C_{pB} + C_{pC} - C_{pA}$ , and  $C_{A0}$  is the concentration of acetone vapor at the inlet. The reaction rate is given as

$$-r_A = k C_{A0} \frac{1 - X}{1 + X} \frac{T_0}{T}$$

### **Flow of a Non-Newtonian Fluid.**

Write a general MATLAB function for solution of a boundary value problem by the shooting method. Apply this function to find the velocity profile of non-Newtonian fluid that is flowing through a circular tube as shown below (Fig. 1). Also calculate the volumetric flow rate of the fluid. The viscosity of this fluid can be described by the Carreau model:

$$\frac{\mu}{\mu_0} = [1 + (t_1 \dot{\gamma})^2]^{(n-1)/2}$$

where  $\mu$  is the viscosity of the fluid,  $\mu_0$  is the zero shear rate viscosity,  $\dot{\gamma}$  is the shear rate,  $t_1$  is the characteristic time, and  $n$  is a dimensionless constant.

The momentum balance for this flow, assuming the tube is very long so that end effect is negligible, results in

$$\frac{d}{dr}(r\tau_{rz}) = -\frac{\Delta P}{L} r \quad (1)$$

Where  $\Delta P/L$  is the pressure drop gradient along the pipe and the shear stress is expressed as

$$\tau_{rz} = -\mu \dot{\gamma} = -\mu \frac{dv_z(r)}{dr}$$

Therefore, Eq. (1) is a second-order ordinary differential equation, which should be solved with the following boundary conditions:

No slip at the wall:  $r = R, v_z = 0$

Symmetry:  $r = 0, \frac{dv_z}{dr} = 0$

The required data for the solution of this problem are:

$$\mu_0 = 102.0 \text{ Pa.s} \quad t_1 = 4.36 \text{ s} \quad n = 0.375 \quad R = 0.1 \text{ m} \quad -\frac{\Delta P}{L} = 20 \text{ kPa/m}$$



**Figure 1**

**Method of Solution:** First we define the following two variables:

Dimensionless distance:  $\eta = r/R$

Dimensionless velocity:  $\phi = v_z v^*$

where  $v^* = (-\Delta P)R^2/L\mu_0$ . Eq. (1) can be expanded and rearranged in its dimensionless form into the following second-order differential equation:

$$\frac{d^2 \phi}{d\eta^2} = - \frac{\frac{1}{\eta} \frac{d\phi}{d\eta} + [1 + \lambda^2 (\frac{d\phi}{d\eta})^2]^{(1-n)/2}}{1 - \frac{(1-n)\lambda^2 (\frac{d\phi}{d\eta})}{[1 + \lambda^2 (\frac{d\phi}{d\eta})^2]}} \quad (2)$$

where  $\lambda = t_1 v^*/R$ .

In order to obtain the canonical form of Eq. (2), we apply the following transformation:

$$y_1 = \frac{d\phi}{d\eta}$$

$$y_2 = \phi$$

The canonical form of Eq. (2) is given as

$$\frac{dy_1}{d\eta} = - \frac{\frac{1}{\eta} y_1 + [1 + \lambda^2 y_1^2]^{(1-n)/2}}{1 - \frac{(1-n)\lambda^2 y_1}{[1 + \lambda^2 y_1^2]}} \quad (3)$$

$$\frac{dy_2}{d\eta} = y_1 \quad (4)$$

The set of nonlinear ordinary differential equations (3) and (4) should be solved with the following boundary conditions:

$$y_1(0) = y_{1,0} = 0 \quad (5)$$

$$y_2(1) = y_{2,f} = 0 \quad (6)$$

The initial value  $y_1(0)$  is known, but the initial value  $y_2(0)$  must be guessed. We designate this guess, in accordance with Eq. (5.120), as follows:

$$y_2(0) = \gamma = \frac{\left[ \frac{(-\Delta P)R^2}{4L\mu_0} \right]}{v^*} = \frac{1}{4} \quad (7)$$

The right-hand side of Eq. (7) corresponds to the velocity of the fluid at the center of the pipe if it was a Newtonian fluid with the viscosity  $\mu_0$ .

The complete set of equations for the solution of this two-point boundary-value problem consists of:

1. The four system equations with their known boundary values [Eqs. (3)-(6)]

2. The guessed initial condition for  $y_2$  [Eq. (7)]
3. Construction of the Jacobian matrix
4. Calculation of the  $\delta y$  vector
5. Correcting the guessed initial conditions.

Once the velocity profile is determined, the flow rate of the fluid can be calculated from the following integral formula:

$$Q = \int_0^R 2\pi r v_z dr$$